Corrections

Vapor–Liquid Equilibrium Data for the Binary Mixture Difluoroethane (HFC-32) + Pentafluoroethane (HFC-125) of an Alternative Refrigerant. Xiaohong Han, Guangming Chen,* Xiaolong Cui, and Qin Wang, *J. Chem. Eng. Data* **2007,** *52,* 2112–2116.

Page 2112. There is one typographical error in the Title: Difluoroethane (HFC-32) should be Difluoromethane (HFC-32).

Also, there is one error in the Abstract: difluoroethane (HFC-32) should be difluoromethane (HFC-32).

Also, there are two typographical errors in the Introduction: chlorodifluoroethane (HCFC-22) should be chlorodifluoromethane (HCFC-22) and difluoroethane (HFC-32) should be difluoromethane (HFC-32).

Page 2113. There is one typographical error in the Experiment section: Difluoroethane (HFC-32) should be Difluoromethane (HFC-32).

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Density, Viscosity, Refractive Index, Excess Molar Volume, Viscosity, and Refractive Index Deviations and Their Correlations for the (Formamide + Water) System. Isobaric (Vapor + Liquid) Equilibrium at 2.5 kPa. Viviana Campos, Ana C. Gómez Marigliano, and Horacio N. Sólimo,* *J. Chem. Eng. Data* **2008,** *53,* 211–216.

Page 214. There is one typographical error in Table 6. The boiling temperature of pure formamide ($x_1 = 1.000$) should be 384.8 K instead of 359.5 K. Additionally, the results were not accurate enough to derive activity coefficients. Therefore, the activity coefficients previously reported in this table should not be considered. Consequently, Table 6 should be as below.

Table 6. Experimental Results for the Mole Fraction of Formamide in the Liquid x_1 and Vapor y_1 Equilibrium Phases and Temperature T for {x_1 Formamide + (1 − x_1) Water} at P = 2.5 kPa

x_1	y_1	T/K
0.000	0.000	279.4
0.232	0.004	293.55
0.391	0.004	302.05
0.466	0.005	305.97
0.545	0.002	310.05
0.589	0.002	314.82
0.748	0.004	320.45
0.806	0.025	330.75
0.879	0.076	339.05
0.937	0.364	352.75
1.000	1.000	384.8

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